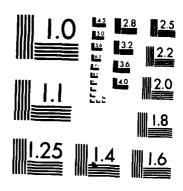
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USAFOEHL REPORT

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DOVER AFB CHARACTERIZATION/HAZARDOUS WASTE MANAGEMENT SURVEY, DOVER AFB DE

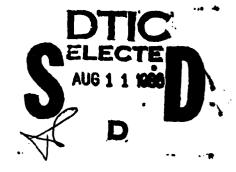
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July 1986

Final Report



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USAF Occupational and Environmental Health Laboratory
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This report has been reviewed and is approved for publication.

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into the sanitary sewer. 34) Perform periodic EP Toxicity testing on neutralized battery aci 55) Repipe vats in building 719 to provide piping dedicated to each vat. 36) Negotiate with solvent recovery representative to provide a system for a trial period.

ACKNOWLEDGMENTS

The authors would like to express their appreciation for the support of lLt Robert A. Tetla, Consultant, MSgt Horace Burbage, SrA Tammy Johnson, AlC Pete Davis, and AlC Ross Simmons, technicians, USAFOEHL/ECQ, in accomplishing this survey. The support of Capt Link Waterhouse, MSgt Hartman, and the other members of the Dover Environmental Engineering Section was greatly appreciated as well.

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I. INTRODUCTION

In a 6 Nov 85 letter, the USAF Hospital Dover, Bioenvironmental Engineering Section (SGPB) requested the USAF Occupational and Environmental Health Laboratory Environmental Quality Branch (USAFOEHL/ECQ) conduct a survey to quantify the wastewater contaminants in the industrial sewer system and make recommendations for pretreatment control (Atch 1).

The survey was conducted by Majors Robert D. Binovi and Elliot K. Ng, lLt Robert A. Tetla, 2Lt Francis E. Slavich, MSgt Horace C. Burbage, Sgt Tammy W. Johnson, and AlCs Robert P. Davis and Ross W. Simmons, USAFOEHL/ECQ from 25 February to 7 March 86.

II. BACKGROUND

A. Introduction

Dover AFB, the home of the 436th Military Airlift Wing since 1966, is located a few miles south of the city of Dover, Kent County, Delaware. The base serves as a large military cargo terminal and supports large transport aircraft. Base population at the time of the survey was approximately 15,366 people.

Kent County has a continental type of climate, with well defined seasons. The Atlantic Ocean, Delaware Bay, and Chesapeake Bay exert considerable modifying influence on the climate. The warmest period of the year is the last part of July, when the maximum afternoor temperature averages 89 degrees F. The coldest part of the year is the last weeks in January and the beginning of February, when the early morning temperatures average near 24 degrees F. The average high and low temperatures for the survey period were 39.5 and 27.7 degrees F.

The average annual precipitation for Dover is 46 inches. The monthly distribution is fairly uniform during the year; August being the wettest month. The precipitation during the survey period 25 Feb-7 Mar totaled .06 inches.

Industrial operations stem from facility, aircraft, and vehicle maintenance. Wastewater from most of the industrial operations flows through a separate sewer system from the sanitary system. Industrial wastewater is pumped from a lift station (Site 4) into the sanitary system. The combined sanitary and industrial wastewater stream flows off the base property to the Kent County Publicly Owned Treatment Works (POTW) by way of Kent County Lift Station 6. The base industrial treatment system previously had two lagoons to facilitate sedimentation, separation, and oxidation of the industrial wastes. Due to groundwater contamination considerations, the lagoons now have been drained and the flow bypassed.

B. Description of Facilities and Industrial Activities

General. The industrial sewer system is designed to collect the effluent from eight buildings and an open wash rack. Attachment 1 summarizes the chemicals used by each industrial operation during the survey. A description of the operations conducted in each building is as follows:

- a. Aircraft Wash Rack, Building 706. This facility is used for washing aircraft and other large parts. During the survey, two aircraft were washed by Galaxy Corp., the corrosion control contractor. Galaxy uses an approved biodegradable aircraft detergent, Calla 800, manufactured by Calla Chemical Operation, P.O. Box H, Stanton CA 90680, for aircraft washing. A petroleum distillate solvent, PD-680 Type II is used also. The amount of chemicals depends upon how long since the aircraft was washed last. According to Mr Nguyen, the Galaxy Supervisor, approximately half a drum of Calla 800 and 30 to 35 gallons of PD-680 Type II are used per aircraft.
- b. Open Wash Rack, Adjacent to Building 582. This is an outside wash rack located on the ramp next to building 582. It serves the same function as building 706. Due to the cold weather, this facility was not used during this survey.
- c. Paint Stripping Facility, Building 582. This building is primarily used for aircraft paint stripping but also houses the administrative office of Galaxy Corp. The building is used jointly by Galaxy, the Field Maintenance Squadron's Jet Engine and Non-powered AGE shops and Aerial Port Squadron. During the time of the survey, all paint stripping activities in this building had ceased. Operations will resume after drain screens are in place. The residues from the floor drain screens will be drummed and brought to the Defense Reutilization Management Office (DRMO) for disposal. The only operation conducted during this period was cleaning Aerial Port forklifts, using one quart of 815-MX aircraft detergent.
- d. Jet Engine Shop, Building 725. This shop is connected to the industrial sewer system by a single floor drain. The shop supervisor indicated no chemicals are disposed of through it, except the detergent used to clean floors.
- e. Jet Engine Shop, Building 719. There are five separate shops in this aircraft maintenance facility, the Cleaning Room, the Components Repair Shop, the GTU Shop, Non-powered AGE Propulsion Shop, and Modules and Accessory Repair Shop. However, the Cleaning Room and the Components Repair Shop are the major industrial waste generators.
- (1) The Cleaning Room is the major industrial activity in this building. The operation includes stripping, degreasing, and descaling of aircraft parts. According to MSgt Lapinski, waste chemicals and sludges are drummed and disposed of through DRMO. However, large amounts of rinse water are used to remove the stripping compounds, etc., and consequently 5-10% of the industrial chemicals along with the residues from the stripping and scraping operation are washed down the industrial sewer system. The shop personnel have had a 3.5-by-11-foot drip pan constructed to collect the paint

- sludge. The contents of this drip pan will be drummed and turned in to DRMO. The shop also cleans aircraft parts by dipping them into vats containing hot carbon remover, PD-680 Type II, and descaling compound. As an interim control measure, the valve draining each vat has been locked to prevent disposal into the industrial sewer system. A project to install a permanent pump and to modify the piping so that shop personnel could empty the contents of the vats directly into 55-gallon drums is planned.
- (2) The Components Repair Shop also generates chemical wastes in building 719. According to MSgt Lapinski, about 10 gallons per month of trichloroethane is used in an ultrasonic cleaner. Spent trichloroethane is drained and disposed of in drums.
- (3) Located alongside the building is a storage area for drums. Drums are dedicated to collecting waste trichloroethane, used oil, JP-4, and PD-680 Type II. A drum is reserved for unknown waste chemical mixtures, to collect mixed chemical wastes from the Propulsion Branch shops.
 - f. Paint and Fiberglass Shop, Building 721.
- (1) This facility was under renovation at the time of the survey. The Fiberglass Shop was temporarily housed in building 720. The Fiberglass Shop operations have little potential for contributing to the industrial wastewater loading. Generally, the small amount of waste generated is in the solid form. This waste is containerized, then taken to the storage area alongside building 719 and placed into drums for disposal.
- (2) The Paint Shop was the only industrial operation in building 721. Methyl Ethyl Ketone (MEK) is used for paint thinning. The spent MEK, along with the remaining lacquer, and other chemicals such as toluene and polyurethane thinner are drummed and disposed of through DRMO. A drum storage area for these wastes is located alongside this building. The shop personnel are attempting to procure a solvent recovery system for MEK, PD-680 Type II, thinners and toluene.
- g. Aircraft Maintenance Shop, Building 724. This building contains the Metal Plating, Welding, and Machine Shops. The shops have minimal impact on the industrial sewer system, as the chemicals used in the processes are either used up (e.g., solder) or are collected and disposed of by contractor (e.g., cyanide plating waste). Metal plating is done in a secured room. Two valves prevent the vats from being accidentally discharged into the industrial sewer system. Machine shop personnel said that they do not discharge any chemicals into the industrial sewer and dispose of about five gallons of PD-680 Type II into the PD-680 drum alongside building 721.
- h. Refueling Vehicle Maintenance Shop, Building 636. Maintenance of the large refueling vehicles that service aircraft is performed in this building. The shop uses PD-680 Type II, a degreaser composed of hydrocarbons in the boiling ranges of normal alkanes with 9 to 14 carbon atoms, and 815-MX detergent to wash floors. Shop personnel estimate that 1-2% of the jet fuel JP-4 enters the industrial sewer system and is collected at the adjacent oil/water separator. The bulk of the waste tankage is collected

and either turned in to Base Supply or used for fire training purposes. Used motor oil and PD=680 Type II are collected in drain pans, drummed and turned in to DRMO. During the survey eight gallons of 815-MX and 47 gallons of JP-4 were used. Two gallons of ethylene glycol were disposed of in the sewer system.

i. Vehicle Maintenance Shop, Building 635

- (1) The Allied Trades and General Purpose Vehicle Maintenance Shop are located in this building. Mr F. Weaver, supervisor for Allied Trades, said nothing is disposed of into the industrial sewer system. However, a small quantity of hydrochloric acid was being disposed of in the sewer, according to the chemical inventory. The hydrochloric acid is brushed on radiators before soldering, which is then washed off with water into the industrial system. Mr Weaver states the quantity of acid is less than a gallon per year. Painting is performed in the Allied Trades Shop. The drain for the paint spray booth waterfall had been sealed off from the industrial sewer system, preventing the waterfall tank from being emptied into the sewer. This will require the wastewater and sludge to be pumped into drums, sampled and disposed of as hazardous waste through DRMO, if sampling indicates it's hazardous. The floor drain connected to the industrial sewer system is covered with paper during painting. Finally, paint stripping is not performed in this building. Waste thinner is placed in drums at the drum storage area, located alongside building 719.
- (2) According to the supervisor of the General Purpose Vehicle Shop, the only significant industrial wastes entering the industrial sewer are wastewater containing 815-MX detergent from floor washing, and neutralized battery acid. Used oils and antifreeze are placed in drums and turned in to DRMO.
- j. Entomology Shop, Building 921. This shop is not connected to the industrial sewer system; however, drains from sinks, etc., are connected to the sanitary sewage system. During the survey period, less than 10 gallons of rinse water were discharged after rinsing the sprayers. Sprayers containing Ficam, Dursban, and Carbamete 15 were rinsed out.
- k. Fuel Cell Repair, Building 945. The hangar drains are connected to the sanitary sewage system through an oil/water separator and explosion-proof lift station. Aircraft fuel tanks are repaired and cleaned at this facility.
- C. Description of Industrial Wastewater Collection System. The following is a description of the industrial wastewater collection system. A schematic is included as Figure 1.
- 1. Site 4, Lift Station at old lagoons, near building 610. Site 4 lift station receives industrial wastewater from a four-inch diameter force main from the lift station servicing the vehicle maintenance compound near building 635 and from a three-inch diameter force main from the lift station

near building 719. The Site 4 lift station is equipped with two 175 gpd pumps, which operate alternately and lift the wastewater to the sanitary sewer through a four-inch diameter force main.

- 2. Site 7, Last manhole on Dover AFB before Kent County Lift Station 6. An 18-inch diameter sanitary sewer transports the combined domestic wastes and industrial waste to the Kent County Pumping Station 6. The wastewater is pumped from here to the Kent County POTW.
- 3. Industrial Separator, Building 583. This large industrial wastewater gravity separator, consisting of twin sedimentation basins and a sludge pit, is housed in building 583. The effluent from the aircraft wash rack at building 706 and the paint stripping facility, building 582, flows to this separator by gravity and is pumped to the lift station near building 719.
- 4. Lift Station, Building 719. The lift station near building 719 accepts flow from the separator at building 583, gravity flow from the Engine Shop at building 725 and a three-inch force main from the lift station near building 724.
- 5. Lift Station, Building 724. The lift station near building 724 accepts gravity flow from the Paint and Fiberglass shop, building 721, and the Aircraft Maintenance shop, building 724.
- 6. Lift Station, Building 635. The lift station in the parking lot of the vehicle maintenance compound receives gravity flow from Refueling Vehicle Maintenance shop, building 636, and Vehicle Maintenance shop, building 635.
 - D. Dover AFB Wastewater Discharge Limitations
- 1. The Delaware Department of Natural Resources and Environmental Control sampled at Site 4 in June 1985. They found a total extractable phenol concentration of 6.64 mg/L and 600 microgram/L chromium. They did not test for methylene chloride. Their results are shown in Attachment 2.
- 2. The base was issued an Industrial Wastewater Discharge Permit (included as Attachment 3) by the Kent County Regional Sewage Disposal District on 1 October 1985 regulating the wastewater discharge from Site 7. The discharge permit was amended on 10 December 1985 to also regulate the discharge from Site 4. In addition, the permit requires the base to reimburse the county for yearly priority pollutant analysis, and incorporate the scheduled base objectives in reducing industrial waste discharge. The effluent parameters, limitations, and monitoring schedules are contained in Table 1.

Table 1

Effluent Limitations and Monitoring Schedule

	Maximum Concentration (m 24 Hour Flow	g/L) Maximum
Effluent Parameter	Proportioned Composite	Instantaneous
Arsenic	0.1	At no time shall
Barium	4.0	the hourly concentration
Cadmium	0.03	of the discharge exceed
Chromium-total	0.5	three times the average
Copper	1.0	concentration.
Lead	1.0	
Mercury	0.01	
Nickel	0.50	
Selenium.	0.50	
Silver	0.2	
Zine	3.00	
Cyanide-total	1.50	
Phenol	4.0	

Monitoring Requirements

Parameter	Frequency	Type sample
COD Phenol Chromium Cadmium Lead Copper Mercury Zinc Oil and Grease	Quarterly	24 hr Composite
EPA Priority Pollutant Scan	Semiannually	24 composite except purgeable organics which will be a grab

III. PROCEDURES

A. Flow

1. Flow from the industrial sewer system was measured at Site 4, the old lagoon lift station sump. Measurements were taken by recording the cycling of the alternating 175 gpm pumps with a Manning 1100XU flow meter. The flow meter was calibrated to record the difference in water elevations in the sump between the high water level pump on elevation as 100%, and the low water level pump off elevation as 0%. The number of cycles could be counted

from the 24 hour recorder and multiplied by the volume of water calculated from the sump dimensions and the elevations of on-off pump operation, to find flow. However, this method neglects the volume of water flowing into the sump while the pump is operating. To correct for this, an alternate method was also used. A determination of the time the pump was off was made by scaling this time from the recorder chart and subtracting from the total time of the measurement, then multiplying by the pump capacity to obtain the slightly higher flow.

2. Total sanitary sewage flow from the base into Kent County Lift Station 6 is normally measured at the lift station by an ultrasonic flow meter. However, the flow meter wasn't operating during the survey period, and hadn't since late December. Consequently, daily flow from the base could not be obtained. Kent County reported the average daily flow for March and April 1986 was 1.1 and 1.13 million gallons per day.

B. Sampling

Cita Number

1. Sampling Site Numbers and Locations. A list of sampling site numbers and locations where the samples were taken is shown in Table 2.

Table 2

Sampling Site Locations

Site Location

Sice Mulio	et. Sice pocacion
GN86006	6" gravity outfall from bldg 706 into bldg 583 separator
GN86008	6" gravity outfall from bldg 582 into bldg 583 separator
CN86009	6" gravity outfall from bldg 725 into lift station near bldg 719
GN860010	6" gravity outfall from bldg 719 into lift station near bldg 719
GN860011	6" gravity outfall from bldg 721 into lift station near bldg 724
GN860012	6" gravity outfall from bldg 724 into lift station near bldg 724
GN860013	6" gravity outfall from bldg 636 into o/w separator near bldg 636
GN860014	lift station sump near bldg 635
GN860015	Site 4, old lagoon lift station sump
GN860016	Site 7, manhole 60 Lebanon last AF manhole before Kent County
•	lift station 6
GN860017	service from bldg 921 in manhole 413, near Entomology bldg 921
GN860018	o/w separator sump near bldg 945

2. Sampling Frequency. Seven days of 24 hour samples composited hourly were taken at sites GN860015 and GN860016. Sampling at other sites was performed over a 24 hour period, with samples composited hourly. Sampling at GN860016 was composited proportionate to typical flow obtained from historical flow data from the Kent County Lift Station 6 flow meter. December 5, 1985 was selected as representative of typical flow. Composite samples were collected with Isco Model 2100 Automatic Wastewater Composite Samplers. Also, daily grab samples were collected for those analyses requiring this type of collection. Samples were analyzed for the parameters listed in Table 3.

Table 3
Sample Analysis

Analysis	Preservation	EPA Method	Where	<u>Who</u>
Biochemical Oxygen Demand	none	405.1	on-site	USAFOEHL
Chemical Oxygen Demand	H ₂ SO ₄ , 4DRGC	Hach Mod. 410.4	on-site	USAFOEHL
Kjeldahl Nitrogen	H ₂ SO ₄ , 4DRGC	305.?	off site	Biospherics
Total Rec. Oils & Grease	H ₂ SO ₄ , 4DRGC	413.?	n	н
Total Organic Carbon	M II	415.?	п	n
Total Cyanide	NaOH, "	335.?	n	п
Total Rec. Phenolics	H ₂ SO ₄ , "	420.?	n	п
As, Ba, Cd, Ca, Cr, Cu, Fe, Pb, Mg, Mn, Ni, Se, Ag, Zn	HNO ₃ "	200.7	R	•
Acid/Base/Neutral Extractables	Ħ	625	Ħ	n
Purgeable Organics	H ₂ SO ₄ "	624	*	n
Residue, Nonfilterable	none	160.2	on-site	USAFOEHL

C. Pretreatment Study. Jar testing was performed on samples of industrial wastewater taken from Site 4 to determine the effectiveness of coagulation-flocculation pretreatment. The Phipps Bird apparatus was used with 2000 mL beakers. The procedure for the jar tests included a one minute 100 rpm rapid mixing cycle followed by 30 minute flocculation at 20 rpm and finally one hour quiescent settling. Samples were obtained by carefully decanting from the beakers.

D. Hazardous Waste Survey.

1. Visits to each building on the industrial sewer system were made to observe the industrial activities and to discuss industrial waste disposal practices with shop personnel. Supervisors were asked to account for the chemical usage during the survey period by recording it on a survey form. In some shops, it was determined that accounting was unnecessary because of minimal chemical usage. Other shop supervisors had difficulty determining daily usage but were able to account for chemicals over weekly or monthly periods.

2. The survey included obtaining an updated list of chemicals, determining the quantity used, and the disposal method for each chemical. This information is contained in Attachment 14.

IV. RESULTS AND DISCUSSIONS

A. Flow measurements. Flow measurements from the lift station at Site 4 are contained in Attachment 4. Twenty-four hour flow measurements were hindered by the shortened battery life due to the cold temperatures encountered during the survey. Flows ranged from 18,000 to 35,000 gallons per day. This amounts to less than 3% of total sewage flow.

B. Wastewater Characterization

- 1. Metals (As,Ba,Ca,Cd,Cu,Fe,Pb,Mn,Hg,Ni,Se,Ag,Zn,Mg). The results of the sampling for metals are contained in Attachment 5. Metals can enter the sewage treatment system from processes such as corrosion control, plating, aircraft and vehicle washing, and battery maintenance. Results showed the vehicle maintenance sump building 635 contained water which exceeded the limits for cadmium, copper, lead, mercury, and zinc. Battery acid disposal and/or corrosion control sludge disposal is probably responsible. Chromium and cadmium concentrations exceeding the limits were found in the outfall from building 719. Cadmium concentrations were exceeded at the outfall from building 706 and at Site 4. Contributions to Site 4 exceeding the cadmium limit are coming from buildings 719 and 635.
- 2. Cyanides. The results of sampling for cyanides are contained in Attachment 6. Cyanides in wastewater are normally associated with plating wastes. Cyanides were not found at any of the sampling sites.
- 3. Phenols. The results of sampling for phenols are contained in Attachment 6. Phenols in the wastewater are normally associated with phenolic based paint strippers, an alternative to methylene chloride paint strippers. Results from all sites were below the 4.0 mg/L limit. A residual of 1.37 mg/L was found from building 582, the corrosion control facility. Although phenolic paint strippers are not used currently, residues may still remain in the drains to building 582.
- 4. Purgeable Organics. The results of sampling for the EPA Method 624 Purgeable Organics are included in Attachment 7. A list of purgeable organics tested for is included as Attachment 8. No limits have been established on purgeable organics. Detectable concentrations of toluene and ethyl benzene were found in the sample from the lift station at building 635. Detectable concentrations of benzene and toluene were found in the oil/water separator at building 945 and a small amount of trichloroethane was found in the manhole at the Entomology Shop, building 413. Significant amounts of methylene chloride, and 1,1,1 trichloroethane were found in one or more days at Site 4, with lesser amounts of trichloroethylene, toluene, and chloroform. These were not detected at any other operation, and appear to be attributable to either sludge or residual water in the Site 4 lift station sump. No purgeable organics were found in any of the seven day grab sampling at Site 7, where the combined industrial and sanitary sewage leave the base.

- 5. Base/Neutrals and Acids. The results of sampling for the EPA Method 625 Base/Neutrals and Acid Extraction are included in Attachment 10. A list of base/neutral and acid organic compounds tested for is given in Attachments 8 and 9. Concentrations of phthalates commonly associated with the washing of JP-4 residue, were found at the aircraft wash rack, Jet Engine Shop, and Refueling Vehicle Maintenance. The phthalates probably are being formed from the oxidation of the xylenes contained in the fuel. This method identified the phenols that were found at the outfall of building 582, corrosion control facility, probably from residuals of phenolic paint strippers. Pentachlorophenols are used as a wood preservative in the Allied Trades shop.
- 6. Five Day Biochemical Oxygen Demand (BOD-5) and Chemical Oxygen Demand (COD)
- a. BOD=5 results are included in Attachment 11. The average value of BODs taken daily over the seven day sampling period at Site 4 and Pumping Station 6 were 176 and 86 mg/L, respectively. High BOD=5 concentrations were found in the sump of the lift station near building 635 (36,502 mg/L) and from building 719 (925 mg/L).
- b. COD results are included in Attachment 11. The average COD concentration taken daily over the seven day sampling period at Site 4 and Pumping Station 6 were 659 and 369 mg/L, respectively. High COD concentrations were found at the sump of the lift station near building 635 (86,000 mg/L), and building 719 (1,600 mg/L).
- c. The ratio of BOD/COD for Site 4 and Pumping Station 6 is .27 and .23, respectively. The ratio of BOD/COD for buildings 635 and 719 is .42 and .58, respectively. Higher ratios indicate wastewater with contaminants more readily biodegradable or less inhibitory to biodegradation. Buildings 635 and 719 had high concentrations of oils and grease (see below) which are generally more readily biodegradable.
- 7. Total Recoverable Oils and Grease. Industrial operations contributing to the oils and grease concentrations at Site 4 appear to be buildings 635 and 636 (lift station at 635 was full of oil), and building 719. The concentration of oils and grease (35,5 mg/L avg) at Pumping Station 6 fall is in the low to normal range for domestic sewage. Results of the sampling for oils and grease are given in Attachment 6.
- 8. Nonfilterable Residue (Suspended Solids). Results of nonfilterable residue sampling are presented in Attachment 11. The average concentrations at Site 4 and Pumping Station 6 are 29 and 83 mg/L, respectively.
- 9. Total Organic Carbon, Total Kjeldahl Nitrogen (TKN). Results of sampling for these parameters are included in Attachment 6. Total organic carbon concentrations ranged from 16 mg/L from building 582 to 440 mg/L from building 719. TKN values ranged from 0.2 mg/L from building 725 to 77 mg/L from buildings 719 and 921.

C. Pretreatment Jar Tests. These very preliminary studies of the wastewater from Lift Station 4 showed alum flocculation-coagulation treatment effective for the removal of cadmium and chromium without pH adjustment. Over 90% of the cadmium and chromium were removed at pH values 7.03 and 7.67. Significantly reduced removal efficiency (66% Cd, 52% Cr) was observed when ferric chloride was used as a coagulant. The optimum dosage of alum appears to be in the 130 to 150 mg/L range; thus, approximately 40 pounds of alum per day would be required. Results of the metal and suspended solids removal efficiencies are shown in Attachment 12.

V. OBSERVATIONS AND CONCLUSIONS

- A. From a review of the previous sampling results and the imposed Kent County effluent standards, corrosion control operations (excluding washing) are mainly responsible for the base exceeding limitations at Site 4, especially for the parameters cadmium, chromium, and phenols.
- B. Levels of chromium and phenols were significantly lower than the State's sampling results of June 1985 and within present and projected levels for these parameters. The fact that no paint stripping at building 582 took place during the survey may account for this. However, the cadmium limit of 0.03 mg/L was exceeded at Site 4 each day during the seven day sampling period. The cadmium found at building 706 may have originated from abrasive or acid cleaning of aircraft parts. The cadmium found in the effluent from building 635 may be traced to paint sludge or solder used for automotive repair.
- C. Effective cadmium and chromium removal is possible by gravity sedimentation with aluminum sulfate addition. This process results in the generation of significant quantities of sludge, with solid waste concentrations of cadmium and chromium possibly exceeding their respective 1 mg/L and 5 mg/L limits for hazardous waste under 40 CFR 261. Cadmium and chromium removal can also be carried out with gravity sedimentation after addition of lime or sulfites. The soluble hexavalent chromium ion (chromate, chromic acid) needs to be reduced to the insoluble trivalent ion (chromium oxide, chromic hydroxide) to facilitate effective precipitation. The good removal efficiency seen in the jar tests indicates the chromium may already be in the trivalent oxidation state, possibly reduced by the relatively high iron concentration found in the water. Sufficient alkalinity was present to preclude lime addition. Additionally, if the waste fails the EP toxicity test for chromium alone, the waste may be excluded from being a hazardous waste, if the chromium is primarily in the trivalent ionic state.
- D. Kent County, by regulating the discharge at Site 4, and at Pump Station 6 has, in fact, imposed stringent pretreatment standards on the base since they do not consider the sizable dilution of the domestic wastewater as partial or complete substitute for adequate treatment. The 10 December 85 Kent County letter (Atch 4) to the base states that Dover AFB does not fall into an EPA categorical standard and therefore the Federal priority pollutant limitations do not apply. Their concern at a local level is based on three reasons:

- 1. Toxicity testing may soon become part of NPDES permits.
- 2. County personnel working at Pumping Station 6 may be subjected to fumes emanating from the wastewater channel which may contain priority pollutants.
- 3. Base personnel maintaining industrial lift stations or working downstream of the base industrial system may be subjected to fumes from the wastewater.
- a. Using Henry's Law to calculate volatile organic concentrations from the wastewater channel from their vapor pressures, neither concentrations nor exposure times would be great enough to be a significant health problem for either county or base personnel (See Attachment 13).
- b. Aquatic toxicity testing using both vertebrates and invertebrates is being included more frequently as part of NPDES permit monitoring for Air Force wastewater treatment plants. It is conceivable that Kent County would impose this type of requirement on the Site 4 discharge. The effects of the cadmium and chromium concentrations at Site 4 would probably be seen if the NPDES toxicity testing of the effluent were performed using the invertebrate Daphnia, as Daphnids are more sensitive to these metals than the organics. Cadmium and chromium concentrations of 24-118 and 455 micrograms/liter, respectively, have been shown to be toxic to 50% of the test organisms (LC50). Acute toxicity to various chemicals is given in Attachment 15.
- c. Whereas the base would most probably meet their effluent standards at Pump Station 6, unless pretreatment for cadmium and chromium or change in waste collection practices take place, the base probably will fail to meet the imposed pretreatment standard at Site 4.
- E. Leaching from the oil and sludge left in the lift station sumps contribute to the daily pollution loading by increasing concentrations of metals, oils and grease, and methylene chloride at Site 4.
- F. The base has an active hazardous waste program. All shop personnel contacted appeared to be acutely aware of the importance of proper disposal and containment of chemical waste. Obviously, there have been many recent procedural and engineering changes preventing conscious and unconscious disposal of chemical waste into the industrial sewer system. Mr Witmer, Base Environmental Coordinator, and Capt Waterhouse, Base Bioenvironmental Engineer, are aggressively seeking every opportunity to reduce the industrial waste discharged into the industrial sewer system.
- G. Since a large portion of the waste streams are spent solvents, the base is trying to procure two solvent recovery systems for MEK, PD-680 Type II, thinners, and toluene. The system being considered is the RX-35 System manufactured by the Recyclene Products, Inc., 1910 Trade Zone Blvd., San Jose, California 95131, (408) 945-8600.

H. Drip pans are available or are being constructed to reduce paint stripping wastes from entering the industrial sewer system. If used conscientiously will help reduce the levels of cadmium, chromium, and methylene chloride from entering the sewer.

VI. RECOMMENDATIONS

- A. Remove the sludge from lift stations at buildings 719 and Site 4, and the oil and sludge from building 635 lift station. Routine sampling results should be representative of current conditions not an indication of past disposal practice.
- B. The base should explore substitution of Plastic Media Blasting (PMB) for chemical paint strippers for operations where alternate paint stripping methods are permitted. In demonstrations at Hill AFB, 95% of the media is reused, 5% is disposed of as a hazardous waste. Project officer at Hill AFB is Tom Bwers, AV 458-3534.
- C. Install a pretreatment process for the removal of cadmium and chromium at Site 4, since corrosion control operation effluent from various locations on base combines at Site 4. If the proper process is selected, ancillary removal of phenols, phthalates and volatiles, can be expected. For example, sedimentation with chemical addition (alum) has been reported to remove >90% of the phenol, >88% of the methylene chloride, and >94% of the di-n-butyl phthalates while removing >98% chromium and >88% of the cadmium in a full scale operation at a paint manufacturing plant.
- D. Battery acid from lead-acid batteries is currently neutralized and disposed of directly into the industrial sewer system at building 636. A periodic EP toxicity test on the neutralized battery acid is necessary to document that the levels of metal, particularly lead, do not exceed the EP toxic level established by the state hazardous waste program.
- E. A solvent recovery system in theory is attractive and should reduce the quantity of hazardous waste solvents in the long run if properly managed and used. The Navy is already using solvent recovery systems successfully at some of their installations, e.g., the paint shop at the Norfork Naval Shipya: .. However, many solvent recovery systems are commercially available and to evaluate the cost effectiveness of any particular system, e.g., RX-35, based on a desk top study may be presumptuous. Before committing to any particular solvent recovery system, the base should negotiate a trial period with the manufacturer so the efficiency and effectiveness of the system can be properly evaluated. More importantly, the recovered solvents should be analyzed to ensure military specifications are met and are suitable for reuse without restrictions. Finally, the number of solvent recovery systems required may depend on whether plastic media blasting will be used for some paint stripping operations.

F. The planned repiping of the vats in building 719 should include the design of a dedicated piping system for each vat. This would prevent cross-contamination in the barrels caused by residuals left in the pipe. After a baseline characterization is performed, only selected spot check analysis on the drums should be required.

REFERENCES

- 1. APHA. Standard Methods for the Evaluation of Water and Wastewater. 16th ed., Washington, D.C.: American Public Health Association, (1985).
- 2. USEPA. Federal Guidelines: State and Local Pretreatment Programs, EPA 430-9-76-017a, vol. 1, P.E.7, (1977).
- 3. USEPA. Treatability Manual. Technologies for Control/Removal of Pollutants, vol.III, (1980).
- 4. Clark, J.W., W. Viessman, Jr. and M. J. Hammer. Water Supply and Pollution Control. New York: Harper & Row, Publishers, (1977)

Attachment 1 Chemical Usage During Survey

general received when well besides addition assesses.

CHEMICAL USAGE DURING SURVEY

BUILDING 582

SHOP: AERIAL PORT SHOP CONTACT: MR K. BRAGG

(AV 455-6895) DATE USED:

CHEMICAL: DATE USES 815-MX 3 MAR

BUILDING 635

SHOP: ALLIED TRADES SHOP CONTACT: MR F. WEAVER

(AV 455-7222)

CHEMICAL: DATE USED:

6011 WELDING ELECTRODE 24,25,26,27 FEB; 3 MAR

ACID CORE SOLDER 25 FEB; 3 MAR YELLOW BRASS ROD 26,28 FEB ENAMEL THINNER 26 FEB; 4 MAR

ADHESIVE RUBBER 3 MAR
HYDROCHLORIC ACID 25 FEB; 3 MAR
PROPOSAL SOLVENT 24,26 FEB; 4 MAR
LACQUER THINNER 25 FEB; 4 MAR

SHOP: GENERAL PURPOSE SHOP CONTACT: SSGT D. OSTRANDER

VEHICLE (AV 455-6572)

CHEMICAL: DATE USED:
ANTIFREEZE 25,26,27 FEB; 4 MAR

30W OIL 26,27 FEB; 3,4 MAR

10W-30 OIL 25,26 FEB
CLEANING COMPOUND 25,26,27 FEB
WINDSHIELD

SPRAY DEGREASER 25,26,27,28 FEB

GREASE 25,26,27 FEB; 3,4 MAR

CARBURETOR CLEANER 26 FEB; 4 MAR AUTOMATIC TRANSMISSION 25,26 FEB; 4 MAR

BRAKE FLUID 28 FEB; 3 MAR

FLUID

GRG GREASE

BUILDING 636

SHOP: REFUEL VEHICLE SHOP CONTACT: MR J. DWYER

MAINTENANCE (AV 455-6771)

CHEMICAL: DATE USED:

AUTOMOTIVE BRAKE FLUID

10W-30 OIL

3 MAR

24,25,27,28 FEB; 5 MAR

PD-680 28 FEB; 4,5 MAR 30W OIL 24,26,28 FEB; 4,5 MAR

815 MX 24,25,26,27,28 FEB; 3,4,5 MAR

24 FEB

ETHYLENE GLYCOL EMULSION DEGREASER	27 FEB
EMULSION DEGREASER	25,27,28 FEB
JP-4	24,26,27 FEB; 3,4,5 MAR
	BUILDING 719 (CONTINUE)
SHOP: GTU SHOP	SHOP CONTACT: TSGT L. OWREY
Shor: did Shor	(AV 455-6997)
CHEMICAL:	DATE USED:
PENETRATING OIL	24,25,26,27,28 FEB
ASSEMBLE FLUID	24,25,26,27,28 FEB
ANTISEIGE	24,25,26,27,28 FEB
	24.26 FEB
LAYOUT DYE BLUE	25,28 FEB; 1 MAR
RTV SILICON RUBBER	25,28 FEB
ISOPROPYL ALCOHOL	24,25,26,27,28 FEB
XVD-40	24,25,26,28 FEB
MAGNAFLUX CLEANER REMOVER	
RTU 8111	26,27,28 FEB
7808 OIL	26,27 FEB
WHITE PETROLEUM	24,25,26,27,28 FEB
LUBRICANT	
JP-4	24,25,26,27,28 FEB
OIL	24,27 FEB
	BUILDING 721
SHOP: PAINT SHOP	SHOP CONTACT: MSGT J. PERRINE
	(AV 455-6556)
CHEMICAL:	DATE USED:
MEK	24,25,26,27,28 FEB; 1,2 MAR
DOPE AND LACQUER THINNER	24,25,26,27,28 FEB; 1,2 MAR
TOULENE	24,25,26,27,28 FEB; 1,2 MAR
POLYURETHANE THINNER	24,25,26,27,28 FEB; 1,2 MAR
	BUILDING 724
	DOLLD ING TO
SHOP: METAL PLATING AND	SHOP CONTACT: MSGT C. JACKSON
WELDING	(AV 455-6857)
CURMICAL .	NAME HOED.
CHEMICAL: ELECTRODE 6010	DATE USED: 24,25,26,27,28 FEB
ELECTRODE 6010	24,25,26 FEB
SILVER SOLDER	28 FEB
ALUMINUM FILLER ROD	24.25.26,27.28 FEB
TITANIUM	24,25,26,27,28 FEB
. Z . BIT Z VI I	zajezjeojeljeo teb
SHOP: MACHINE SHOP	SHOP CONTACT: MSGT V. WHITE
	(AV 455-6856)
CHEMICAL:	DATE USED:

PD-680 TYPE II

NOTE: SMALL QUANTITY USED ON A CONTINUAL BASIS. TRACKING DAILY USAGE OF CHEMICALS USED IN THIS

SHOP WAS NOT NECESSARY.

BUILDING 725

SHOP: ENGINE SHOP SHOP CONTACT: MSGT S.COOK

(AV 455-6914)

NOTE: CHEMICAL INVENTORY FOR THIS SHOP WAS NOT NECESSARY.

BUILDING 706

SHOP: AIRCRAFT WASHRACK SHOP CONTACT: MR T. NGUYEN

(AV 455-7502)

CHEMICAL: DATE USED:
CALLA 800 SOAP 25 FEB; 1 MAR
PD-680 25 FEB; 1 MAR

י נומיז כב

BUILDING 719

SHOP: CLEANING ROOM SHOP CONTACT: MSGT T. LAPINSKI

(AV 455-6997)

CHEMICAL: DATE USED:

815 MX

NOTE: THESE CHEMICALS ARE USED
DESCALING COMPOUND

IN THE SHOP ON A CONTINUAL
PD-680

BASIS. MSGT LAPINSKI DID NOT

KEROSENE TRACK DAILY USAGE.

EPOXY 3 POLYURETHANE

PAINT REMOVER

SHOP: COMPONENTS REPAIR SHOP CONTACT: MSGT T. LAPINSK1

(AV 455-6997)

CHEMICAL: DATE USED:

COLD CARBON REMOVER NOTE: THESE CHEMICALS ARE USED CALIBRATION FLUID IN THE SHOP. MSGT LAPINSKI DID TRICHLOROETHANE NOT TRACK DAILY USAGE.

TRICHLOROETHANE
EA 934 (PART A)
EA 934 (PART B)
LUBRICANT SOLID FILM
ADHESIVE TYPE I

SHOP: MODULES AND ACCESSORY SHOP CONTACT: MSGT T. LAPINSKI REPAIR (AV 455-6997)

CHEMICAL: DATE USED:

PD-680 TYPE II NOTE: THESE CHEMICALS ARE USED FINGERPRINT REMOVER IN THE SHOP. MSGT LAPINSKI DID

SYNTHETIC ENGINE OIL NOT TRACK DAILY USAGE.

SHOP: NONPOWERED AGE PROPULSION

CHEMICAL:
FLAT BLACK SPRAY PAINT
RED LACQUER SPRAY PAINT
WHITE LACQUER SPRAY PAINT
YELLOW SPRAY PAINT
HYDRAULIC FLUID FIRE
RESISTANT
BRAKE FLUID
GREASE AUTOMOTIVE AND
ARTILLERY
OLIVE DRAB SPRAY PAINT
ALL PURPOSE CLEANER

SHOP CONTACT: MSGT T. LAPINSKI

DATE USED:
NOTE: THESE CHEMICALS ARE USED
IN THE SHOP. MSGT LAPINSKI DID
NOT TRACK DAILY USAGE.

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Attachment 2 Delaware Sampling Results

TECHNICAL SERVICES SECTION

DIVISION OF ENVIRONMENTAL CONTROL

DEPARTMENT OF NATURAL RESOURCES AND ENVIRONMENTAL CONTROL

REQUEST FOR LABORATORY ANALYSIS

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Total mg/1	763	15						
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l, Z-dichloroethene		21.0	41.0	
chloroform		4.2	4.1	
1,2-dichloroethane		1.5	1.3	V
1,1,1-trichloroethane		1200	1400	2.1
trichloroethene		440	650	1.0
tetrachloroethene		98	100	<1.0
lili2.2- tettachloroethane	1	41.0	41.0	41.0
BASE NEUTRAL EXTRACTABLE	s			
bis(2-chloroethyl) ether	uq/1	4100	1600	<10.
bis(2-chloroisoproplyl)		2900	3800	
N-mitroso-d-n- proplyamine		960	1300	
bis(2-chloroethopyl)		2100		
di-N-butyl phthalate		Z 10.	410.	
his(2-ethylhexyl)				
chrysene				
di-n-octylphthalate				
benzo(b)fluoronthene				
pyrene	V	1	V	
ACID EXTRACTABLES				
2-culorophenol	49/1	220	190	< 10-
2-netrophenol	3	130.	110	
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2.4-dimitrophenol		410.	≥10.	
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Attachment 3
DAFB Industrial Wastewater Discharge Permit

Kent



County Engineer

COUNTY ADMINISTRATION
SURDING
414 FEDERAL STREET
DOVER, DELAWARE 19901
SHANDICAPPED ACCESSIOLE
#49-734-2101

December 10. 1985

Base Hospital/SGPB Dover Air Perce Base Dover, Dalaware 19902

Attn: Capt. Lindsey Weterhouse

Ref: DAFS Industrial Westewater Discharge Permit

Gentlemen:

Colonel Richard B. Harper's letter regarding directed actions to be taken to reduce industrial discharges to the Kent County Wastewater Facilities has been reviewed and it is agreed that these are positive steps being taken by the Base to reduce the industrial discharges. We strongly recommend these actions be completely carried out and emphasized with all Base personnel. In order to monitor progress along these lines the enclosed industrial wastewater discharge permit has incorporated a schedule which it is forecasted that the Base will be able to meet if the actions are implemented successfully with Base personnel. Please have the Base Commander sign the permit, make a copy for your file and return the permit to this office by January 1, 1986.

There are three additions to this permit and each will be discussed individually.

- 1. The industrial discharge site is referenced specifically as a designated site as this is the location where all industrial wastes combine and enter the sanitary sewer.
- 2. The Base is required to reimburse the County for a yearly priority pollutant analysis. The County recovers all standard costs incurred in the pretreatment program through County-wide user fees and any extra costs above the standard costs are billed specifically to the industry. The priority pollutant scan is considered an extra cost as the Base was the only contributor determined to be a significant priority pollutant discharger during the testing completed in June, 1985.
- 3. The schedule of compliance incorporates the Bases' objectives in reducing the industrial wastes as previously stated.

Dover Air Force Base

Attn: Capt. Lindsey Waterhouse

Ref:

DAFB Industrial Wastewater

Discharge Permit

December 10, 1985

-2-

The Base is a unique industrial contributor in that a significant amount of priority pollutants have been seen in the Base industrial flow; however, the Base does not fall into an EPA categorical standard and therefore the Federal priority pollutant limitations do not apply. At the local level these pollutants remain a concern for several reasons:

- 1. As the enclosed letter from the Director of EPA Permits Division to NPDES State Directors dated July 24, 1985 indicates, toxicity testing may soon become a part of NPDES permits.
- 2. County personnel working at Pumping Station No. 6 may be subjected to fumes emanating from the wastewater channel which may contain priority pollutants.
- 3. Base personnel maintaining industrial lift stations or working down stream of the Base industrial system may be subjected to fumes from the wastewater.

At the present time there are no local limits for priority pollutants; however, if the above stated reasons necessitate limits, then the appropriate limits would be instated.

We ackowledge that the Base has made recent strides towards reducing the industrial discharges and hope that continued efforts are strongly implemented in these matters.

Very truly yours,

John Wolfenden

Hydraulic Engineer

JW:lm

Eng.

cc: Donald Witmer

William C. Henry, P. E.

KENT COUNTY LEVY COURT

OFFICE OF THE COUNTY ENGINEER

414 FEDERAL ST., DOVER, DELAHARE 19901

INDUSTRIAL WASTEWATER DISCHARGE PERMIT

Code, a	dance with all terms and conditions of the Kent County Sanitary delso with any applicable provisions of Federal or State law or on; permission is hereby granted to:	
_	Department Of The Air Force	
	Dover Air Force Base	
	Dover, Delaware 19902	-
	discharge of industrial wastewater to the Kent County Regional Esposal District at the location designated as	
. •	Cotal DAFB Flow - Kent County Pumping Station #6, Lebanon Road-	
	AFB Industrial Flow - DAFB site #4, industrial wastewater pumping star	tion
This pe	rit is granted in accordance with the application filed on Feb 23, 1985	
the Cou with an	onformity with plans, specifications and other data submitted to ty in support of the above application, all of which are filed considered part of this permit, together with the following named as and requirements.	
	ffective Date: January 1, 1986	
,	xpiration Date: January 1, 1989	
:	ste: Signed	

Permittee, Title

Kent County Engineer

Date: 9 Dec 1985 Signed

Wastewater Discharge Limitations

The discharge from the designated location shall be limited to the effluent quality limitations as defined in Sections 340 - 344 of the Kent County Sanitary Code with the following additions:

		MERCIEUM CON	CERTIFIED .
Effluent Pa	raneter	24 Hour Flow Proportioned Composits	Maximum Instantaneous
Arsenic Berium Cadmium Chrowium	-total	0.1 4.0 0.03	At no time shall the hourly concentration of the discharge exceed three times the average
Copper Lead Marcury		1.0 1.0 0.01	concentration.
Wickel Selenium Silver		0.50 0.50 0.2	•
Zinc Cyanide- Phenol	total	3.00 1.50 4.0	

Monitoring Requirements

The permitted discharge shall be monitored by the permit holder in compliance with the following schedule:

en e		Monitoring Re	equirements
		Measurement	Sample
<u>eff</u>	luent Parameter	Frequency	Type
Industria	al Wastewater Pumping Station	·	•
Site #4	COD	Quarterly _	24 Hr. Composite
	Phenol	Quarterly -	24 Hr. Composite
	Chronium	Quarterly	24 Hr. Composite
•	Cadmium -	Quarterly	24 Hr. Composite
	Lead	Quarterly	24 Hr. Composite
	Copper	Quarterly	24 Hr. Composite
••	Mercury	Quarterly	24 Hr. Composite
	Zinc	Quarterly	24 Hr. Composite
	Oil and Grease	Quarterly	Grab -
	EPA Priority Pollutant Scan	Semi-Annually	24 Hr. Composite except for purgeable organics which will be a grab sample

PERMIT NO. 6

Monitoring Requirements

Cont'd

		Monitoring	Requirements
Effluent Param	eter	Measurement Frequency	Sample Type
Total DAFB Flow	Site #7	· · · · · · · · · · · · · · · · · · ·	
BOD -	÷	Quarterly	· 24 Hr. Composite
TSS '		Quarterly	24 Hr. Composite
Phenol	•	Quarterly	24 Hr. Composite
Chromium		Quarterly	24 Hr. Composite
Cednium		Quarterly	24 Hr. Composite
Load		Quarterly	24 Hr. Composite
Copper		Quarterly	24 Hr. Composite
Hercury		Quarterly	24 Hr. Composite
Zinc		Quarterly	24 Hr. Composite
Oil and Gr	00.00	Quarterly	Crab
pa	ونصي	Quarterly	24 Er. Composite

The above required analyses for site #4 and #7 shall be submitted to the County Engineer's Office on a quarterly basis.

The County Engineer's Office will also complete yearly industrial monitoring as outlined in the Kent County Pretreatment Program. This monitoring normally includes BOD, TSS and heavy metals.

The County monitoring of the Dover Air Force Base industrial discharge will include the normal parameters, however, based upon the sample taken in June, 1985 by this office, a yearly priority pollutant scan will also be completed by this office. The cost of the priority pollutant scan will be billed directly to the Dover Air Force Base and will be itemized on the Dover Air Force Base sewer bill.

•)			
PERMIT NO	6			
•	ses shall be performed references:	ormed in accorda	ance with the lat	est edition of the
	DARD METHODS FOR tion, 1980, America			
Envi	ronmental Protect	ion Agency, Wate	er Quality Office	TES, April, 1971, Analytical Linnati, Ohio, 45268
Schedule	of Compliance			
Apri		OEHL study to ex		priority pollutants
Apri	l 1986 - Limit chr	omium discharge	at site #4 to be	low 0.5 mg/l.
June	1986 - Limit phen	ol discharge at	site #4 to below	4.0 mg/1.
-				
				•
		•		
Rate and	Time of Discharge			
	ge production day shall not exceed		for discharge a	t the designated
The maxim	um hourly	disc	harge flow rate	shall not exceed

PERMIT CONDITIONS

104,000 ent.

General

In consideration of the granting of this permit the undersigned agrees:

To furnish any additional information relating to the installation

PERMIT NO. 6

PERMIT CONDITIONS

Cont'd

or use of the industrial sewer for which this permit is sought as may be requested by the County Engineer.

- 2. To accept and abide by all provisions of the Kent County Sanitary Code and of all other pertinent local laws or regulations that may be adopted in the future.
- 3. To operate and maintain any waste pretreatment facilities, as may be required as a condition of the acceptance into the public sewer of the industrial wastes involved, in an effecient manner at all times, and at no expense to Kent County.
- 4. To cooperate at all times with the County Engineer and his representatives in their inspecting, sampling, and study of the industrial wastes, and any facilities provided for pretreatment.
- 5. To notify the County Engineer immediately in the event of any accident, negligence, or other occurrence that occasions discharge to the public sewers of any wastes or process waters not covered by this permit.

Right of Entry

The permittee shall allow duly authorized employees or representatives of the County to enter the permittee's premises for the purpose of inspection, observation, measurement, sampling, and testing in accordance with Section 300 of the Kent County Sanitary Code.

Sampling Manhole Requirements

If, in the opinion of the County Engineer, there are not adequate facilities for the acquisition of representative samples and accurate flow measurements, the County Engineer can require that a sampling manhole with a flow measuring device be installed by the permittee at his expense. This sampling manhole shall be approved by this office before installation. The permittee shall be responsible for all maintenance of the sampling manhole and calibration of the monitoring equipment.

Change in Wastewater Discharge

All discharges authorized herein shall comply with the terms and conditions of this permit. Any industrial facility expansions, production increases or process modifications which result in new, different or increased discharges of pollutants must be reported by submission of a new industrial waste disposal questionnaire. This permit may be modified to specify and limit any pollutants not previously limited. The discharges of any pollutant more frequently than or at a level in excess of that specified and authorized by this permit shall constitute a violation of the terms and conditions of this permit.

PERMIT	NO.	0	

Permit Modifications

After sufficient notice to the permittee, this permit may be modified, suspended, or revoked in whole or in part during its term for cause including, but not limited to, the following:

- (a) Violation of any terms or conditions of this permit.
- (b) A change in any condition that requires either a temporary or permanent reduction or elimination of the authorized discharge.
- (c) If an effluent standard is established under any State or Federal law for a pollutant which is present in the discharge and such standard or prohibition is more stringent than any limitation for such pollutant in this permit.

Notice of Non-compliance

In the event the permittee does not comply with or will be unable to comply with any daily maximum effluent limitation specified in this permit due to:

- -- (1) Breakdown of industrial wastewater pretreatment equipment.
 - (2) Accidents caused by human error or negligence; or
 - (3) Other causes, such as acts of nature.

The permittee shall notify the operator of the Kent County Wastewater Treatment Plant immediately by telephone so that the operator can take the necessary steps to prevent damage to the wastewater treatment process and equipment. The County Engineer shall be notified in writing within five (5) days and shall include the following pertinent information:

- (1) Cause of non-compliance.
- (2) A description of the non-complying discharge.
- (3) Anticipated time and condition of the non-compliance is expected to continue, or if such condition has been corrected, the duration of the period of non-compliance.
- (4) Steps taken by the permittee to reduce and eliminate the non-complying discharge; and
- (5) Steps to be taken by the permittee to prevent recurrence of the condition of non-compliance.

Nothing in this permit shall be construed to relieve the permittee from the penalties for non-compliance of this permit for any reason subject to the Kent County Sanitary Code.

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Attachment 4 Site #4 Flow

Dover AFB Site 4 Flow

Date	Times	Flow (gal)	
		by Cycles	by Chart time
26 Feb	1430-2400	6,938	7,875
27 Feb	0001-0930	5,204	6,825
28 Feb	0930-2400	13,344	14,175
01 Mar	0001-0500	5,782	6,300
01 Mar	0900-2300	10,407	11,55
02 Mar	1000-2400	10,413	11,02
03 Mar	0900-2400	12,720	14,175
04 Mar	0001-2400	13,298	13,650
05 Mar	0001-0630, 0900-2400	28,910	30,450
06 Mar	0001-2400	23,128	25,725
07 Mar	0001-0900	4,047	5,250

Attachment 5
Sample Sites with Detectable Metal Concentrations

SAMPLE SITES WITH DETECTABLE METAL CONCENTRATIONS AT DOVER AFB

Sample Sites (Conc. in µg/L)

			•	
Substance	0006	0008	0009	0010
Arsenic	3.0	7.0	2.0	41
Barium	25	27	7.0	28
Cadmium	122	21.6	0.4	370
Total Chromium	20	119	6.0	2480
Copper	103	56	8.0	226
Iron	1700	800	2300	2900
Lead	32	206	9.0	410
Manganese	42.4	54	53	46.5
Mercury	<0.2	<0.1	<0.1	<0.1
Nickel	52	<3.0	<3.0	85
Selenium	<1.0	<1:0	<1.0	<1.0
Silver	<0.2	1.0	0.4	2.2
Zine	220	290	30	870
Calcium	25400	20500	19300	22900
Magnesium	7590	6930	6490	7050
				,
	0012	0013	0014	0015
Arsenic	3.0	5.0	<50	6.0
Barium	1210	42	1050	21.4
Cadmium	35.7	17.8	345	211.3
Total Chromium	285	22	400	194.6
Copper	73	58	3350	72.7
Iron	2200	2700	15000	1743
Lead	48	263	14900	88.7
Manganese	34	55.5	335	57 ·
Mercury	0.1	<0.1	20	0.11
Nickel	5.0	<3:0	150	16.6
Selenium	<1.0	<1:0	50	2.1
Silver	2.4	0.7	15	1:1
Zinc	340	440	21500	274
Calcium	29800	24600	106000	23500
Magnesium	14100	8200	31000	6600
			- ,	

	0016	0017	0018
Arsenic	5.3	5.0	6.0
Barium	28.1	82	33
Cadmium	3.3	12.5	18.5
Total Chromium	4:86	7.0	20 ·
Copper	39.4	332	39
Iron	257	16400	1600
Lead	11.7	106	67
Manganese	21.3	209	36.8
Mercury	0.27	0.3	0.1
Nickel	2:1	7.0	6.0
Selenium	2.1	<1.0	<1.0
Silver	3.9	3.5	1.3
Zine	111.4	580	340
Calcium	22200	34200	20400
Magnesium	6850	10700	5700

^{*}Note: Concentrations given for sites 0015 and 0016 are 7-day averages.

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Attachment 6
Sampling Results for Various Parameters

SAMPLING RESULTS FOR VARIOUS PARAMETERS FOR DOVER AFB

Parameter	Sample 0006	Sites(Conc. 0008	in mg/L) 0009	0010
Total Organic Carbon Oil and Grease Total Kjeldahl Nitrogen	143 19.3 2.0	16 9.8 0.5	23 440 5-0 1860 0:2 77	(0.01
Cyanide Phenol	<0.02 .046	<0.01 1.37	<0.01 .07 .095	<0.01
	0012	0013	0014	0015
Total Organic Carbon Oil and Grease Total Kjeldahl Nitrogen Cyanide Phenols	88 6.7 1.9 <0.01	137 408 8.0 <0.01 .137	N/A 184.6 N/A 83.6 N/A 18.6 N/A <.011 N/A .524	
	0016	0017	0018	
Total Organic Carbon Oil and Grease Total Kjeldahl Nitrogen Cyanide Phenol	74.86 35.5 36.1 <0.01 .104	140 67 77 <0.01 .095	25 22.2 16 <0.01 .030	

^{*}Note: Concentrations given for sites 0015 and 0016 are 7-day averages.

Attachment 7
Sample Sites With Detectable Amounts of Purgeable Organics

SAMPLE SITES AT DOVER AFB WITH DETECTABLE AMOUNTS OF PURGEABLE ORGANICS

Site No.	Substance	Concentrations found (µg/L)
0014	Toluene Ethylbenzene	270 31
0015	Methylene chloride	11000, 10000, 21000 6200, 2800, 720
	Tetrachloroethylene Trichloroethylene	230 101, 150, 110
	1,1,1 Trichloroethane	9100, 900, 670
	Toluene Chloroform	120 , 130 170
0017	1,1,1 Trichloroethane	58
0018	Benzene	19
•	Toluene	75

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Attachment 8
Purgeable Organics And Base/Neutral Extractables

PURGEABLE ORGANICS AND BASE/NEUTRAL EXTRACTABLES TESTED FOR AT DOVER AFB

Purgeable Organics

Base/Neutral Extractables

Acrolein Acenaphthylene Acrylonitrile Acenaphthene

Benzene Butyl Benzyl Phthalate
Toluene 1,2-Dichlorobenzene
Ethylbenzene 1,3-Dichlorobenzene

Carbon tetrachloride 1,4-Dichlorobenzene

Chlorobenzene Hexachloroethane

1,2 Dichlorobenzene
1,1,1 Trichloroethane
1,1 Dichloroethylene
1,3 Dichloropropene (cis)
Hexachlorobenzene
1,2,4-Trichlorobenzene
bis (2-Chloroethoxy) methane

Chloroethane Naphthalene

1,1,2 Trichloroethane 2-Chloronaphthalene

1,1,2,2 Tetrachloroethane Isophorone 2- Chloroethyl vinyl ether Nitrobenzene

Chloroform 2,4-Dinitrotoluene

1,2 Dichloropropene 2,6-Dinitrotoluene

1,3 Dichloropropene (trans) 4-Bromophenyl phenyl ether
Methylene chloride bis (2-Ethylhexyl) phthalate

Methyl chloride Di-n-butyl phthalate

Methyl bromide Fluorene
Bromoform Fluoranthene

Dichlorobromomethane Chrysene
Trichlorofluoromethane Pyrene
Chlorodibromomethame Phenanthrene
Tetrachloroethylene Anthracene

Trichloroethylene Benzo(a)anthracene

Vinvl chloride Benzo(b)fluoranthene

1,2-trans-Dichloroethylene Benzo(k)fluoranthene

bis (Chloromethyl) ether Benzo(a)pyrene Indeno(1.2.3-c.d)pyrene

Dibenzo(a,h)anthracene Benzo(g,h,i)perylene

4-Chlorophenyl phenyl ether

3,3-Dichlorobenzidene

Benzidene

bis(2-Chloroethyl) ether 1,2-Diphenylhydrazine Hexachlorocyclopentadiene N-Nitrosodiphenylamine N-Nitrosodimethylamine N-Nitrosodi-n-propylamine Attachment 9
Organochlorine Pesticides, PCBs, and Extractables

ORGANOCHLORINE PESTICIDES, PCBs, AND ACID EXTRACTABLES TESTED FOR AT DOVER AFB

Organochlorine Pesticides and PCBs

Acid Extractables

_			_		_
aln	ha-	·Enc	โดรเ	มโร	'an

Phenol

beta-Endosulfan 2-Nitrophenol

4-Nitrophenol

Endosulfan sulfate

alpha-BHC beta-BHC delta-BHC gamma-BHC Aldrin

Dieldrin

4.4-DDE

4.4-DDD

2,4-Dinitrophenol
4,6-Dinitro-o cresol
Pentachlorophenol
p-Chloro-M-Cresol
2-Chlorophenol
2,4-Dichlorophenol
2,4-Crichlorophenol
2,4-Dimethylphenol
2,4,5-Trichlorophenol

4,4-DDT 2,4,5-Trichlord
Endrin 2-Methylphenol
Endrin aldehyde 4-Methylphenol
Heptachlor Benzoic Acid

Heptachlor epoxide

Chlordane Toxaphene Arochlor 1016 Arochlor 1221 Arochlor 1232 Arochlor 1242 Arochlor 1248 Arochlor 1254 Arochlor 1260 Attachment 10
Sample Sites with Detectable Amounts of Base/Neutral and Acid Extractables

SAMPLE SITES AT DOVER AFB WITH DETECTABLE AMOUNTS OF BASE/NEUTRAL AND ACID EXTRACTABLES

Site No.	Substance	Concentrations found($\mu g/L$)
0006	bis (2-Ethylhexyl) phthalate	280
0008	2-Chlorophenol 2,4,6-Trichlorophenol 2,4,5-Trichlorophenol	1 30 480 360
0009	Di-n-butyl phthalate	29
0012	Di-n-butyl phthaiste 4-Methylphenol	24 35
0015	2,4,6-Trichlorophenol 2,4-Dimethylphenol Di-n-butyl phthalate 2,4,5-Trichlorophenol 1,2-Dichlorobenzene bis (-Ethylhexyl) phthalate	150, 170, 53, 31 130 33, 45 140, }8, 30 22 201
0016	Di-n-butyl phthalate 4-Methylphenol Diethyl phthalate Phenol	44, 120, 98, 94 20, 18, 23, 56, 22 11, 16, 14 39

Attachment 11
Results for pH, Temperature, COD, Suspended Solids, and BOD

DOVER AFB RESULTS FOR pH, TEMPERATURE, CHEMICAL OXYGEN DEMAND (COD), SUSPENDED SOLIDS, AND BIOCHEMICAL OXYGEN DEMAND (BOD)

Samp	Date	рН	Temp(C)	COD mg/L	SS mg/L	BOD mg/L
0006	26 Feb	7.31	20.2	350 20	119	
0008	28 Feb	8.01	21.7	80 9.0	NR	
0009	5 Mar	7.80	19.3	200 18	18.9	
0010	4 Mar	9.61	16.4	1600°	248	925
0012	4 Mar	7.98	19.2	340 29	60.6	
0013	1 Mar	8.57	12.7	550 119	84.1	
0014	3 Mar	NR	13.2	86000	NR	36502
0015	26 Feb	8.09	11.1	840 44	252	• • -
0015	27 Feb	8.41	18.4	820 37	314.3	
0015	28 Feb	8.25	11.9	720 31	337	
0015	1 Mar	7.48	11.0	480 12	218.6	
0015	2 Mar	7.43	9:1	500 19	28	
0015	3 Mar	7.57	7.6	550 32	61.6	
0015	4 Mar	7:98	14.8	700 26	127.5	
0015	5 Mar	7.6	5.4	NR 29	129	
0016	26 Feb	8:03	14.8	349 204	NR	
0016	27 Feb	NR	NR	325 NR	57.7	
0016	28 Feb	7.70	16.1	300 92	84 ·	
0016	1 Mar	7.82	13.1	320 36	83	
0016	2 Mar	7.75	10.4	290 33	83.5	
0016	3 Mar	7.80	10.3	680 57	64	
0016	4 Mar	7.81	14.8	340 106	84.5	
0016	5 Mar	7.84	18.3	350 53	1455	
0017	27 Feb	8.43	16.2	300 342	76.5	
0018	28 Feb	7.72	13.6	100 25	32.5	

Attachment 12 Coagulation and Sedimentation Test

DOVER AFB COAGULATION AND SEDIMENTATION TEST

		(150 mg/L)	(200 mg/	L)	
Metals	0019 (control)	0020	*Reduction	0021	\$ Reduction
рH	7.67				
Arsenic	2.0 µg/L	<1.0	50 %	1.0	50 %
Cadmium	19.9	0.7	96.5\$	1.0	95%.
Chromium	32.0	1.0	96.9%	2.0	93.75% .
Copper	39.0	16.0	59\$	103	-264%
Lead	32.0	11.0	65.6%	12.0	62.5%
Mercury	<0.4	same	N/A	same	N/A
Nickel	29.0	25.0	13.8%	833	-2872 %
Selenium	<1.0	1.0	N/A	1.0	N/A
Silver	0.5	0.4	20%	0:4	20\$
Zinc	190	70.0	63.2\$	80.0	57.9%
Antimony	<.002mg/L	Same	N/A	same	N/A
Beryllium	<.0001 mg/L	same	N/A	same	N/A
Thallium	<.006 mg/L	same	N/A	same	N/A
		(130 mg/L)			
	0022 (control)	0023	#Reduction		
рН	7.03				
Arsenic	4:0	2.0	50\$		
Cadmium	52.8	3.9	92.6\$		
Chromium	42.0	2:0	95.2%		
Copper	54:0	12.0	77.8%		
Lead	49.0	21.0	57.2%		
Mercury	<0.2	same	N/A		
Nickel	46.0	27.0	41.3\$		
Selenium	2.0	1.0	50\$		
Silver	7.6	0.3	96.1%		
Zinc	210	130	38.1\$		
Antimony	<.002 mg/L	same	N/A		
Beryllium	<.0001	same	N/A		
Thallium	<.006	same	N/A		
		(200 mg/L)			
	0024 (control)	0025	#Reduction		
Arsenic	3.0	1.0	66.7%		
Cadmium	85.2	28.6	66.4%		
Chromium	346	165	52.3%		
Copper	86.0	61.0	29.1%		
Lead	67.0	44.0	34.3%		
Mercury	<0.2	same	N/A		
Nickel	17.0	61.0	-359%		
Selenium	1.0	1.0	0%		
Silver	0.7	0.3	57.2\$		
Zinc	320	300	6.2 \$		
•	.002	same	N/A		
Beryllium <		same	N/A		
Thallium <	.006	same	N/A		

DOVER COAGULATION AND SEDIMENTATION STUDY SUSPENDED SOLIDS RESULTS

Sample No.	Alum Conc.(mg/L)	SS Conc.(mg/L)	\$Reduction	
0019	control	10.0	74\$	
0020	150.0	2.6	70 %	
0021	200.0	3.0		
0022	control	36.0		
			100	*
0023	130.0	0.0		
0024	control	63.0	99.6\$	
0025	200.0	0.27		

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Attachment 13
Calculations For Methylene Chloride and 1,1,1 TCE Vapors

ATMOSPHERIC CONCENTRATION CALCULATIONS FOR METHYLENE CHLORIDE AND 1.1.1 TRICHLOROETHANE

From Purgeable Organic Results for 2/28/86:

1,1,1 Trichloroethane = 9.1 mg/L Methylene Chloride = 21.0 mg/L

Molarity:

Methylene Chloride-- .021 gr/1/84.94 gr/mole = 2.47E-4 moles/L

1,1,1 Trichloroethane-- .0091 gr/L/133.4 gr/mole = 6.82E-5 moles/L

Partial Pressures:

MC-- 2.47E-4 moles/1 * 3.19E-3 atm $m^{**}3/mole$ * 1000L/ $m^{**}3$

= 7.88E-4 atm

1,1,1-- 6.82E-5 moles/1 # 4.92E-3 atm m**3/mole * 1000L/m**3

= 3.36E-4 atm

**Note- Second term in previous two calculations is the Henry's Constant for the particular substance

Now for:

MC--7.88E-4 atm * 760 torr/atm = .599 torr

1,1,1--3.36E-4 atm * 760 torr/atm = .255 torr

TOTAL = .599 + .255 = .854 torr

1 of Total Pressure:

= .854 torr/760 torr * 100 = .11% = 1100 ppm

Attachment 14
Hazardous Waste Management Survey Forms

					DOVER AFD	2			PAGE	PAGE 1 OF 3	
SHOP HAME, AE	. 3	P Pent	Arrial Pont		BUILDING, 582	582				24 FEP 5'5	
SHOP SUPERVISOR:	7	Ar Bar	60 ES	- 1				¥ .	455-6	3528	
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1						DOVER AFB				PAGE	PAGE 1 OF 2		
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1169-00-00-011	2		102	202		2/2 02			0350	5	
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RTU 9891 CATALVST 8030-00-142-9128	7										
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SHOP SUPERVISOR		SUBSTANCE	WHITE PETROLEUM LUDRICANT		5P4 Jet Fol		11.2-1.010 0:1			•					٠			

IS ANYTHING RINSED DR COOLED OFF (V/N) 1 00 2 WASTE DISPOSAL METHOD I.e drein DATE u BUTLDING: 725 AMOUNT USED/DAY DOVER AFE 3 2 SHOP NAME: FIBERGLASS SHOP 80/000 63 8393 ---507000 973 53415 pm 2 2 2 2 3 | 2 | 3 | 3 \$ | \$ 16.8h 566 00 0189 85 00 99231SE 223 210 da olos SHOP SUPERVISOR! 1751 EPOXY ADHESIVE 8040-00-959-1854 SEALTING COMPOUND 8030-00-470-9134 9309 84 600- ADHESIVE PuryESTER 8160 828 E193411A SUBSTANCE 1/10

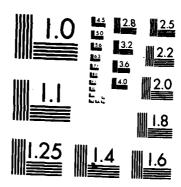
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SHOP SUPERVISOR, MSLT PAKELLE JAMOS	3	Lr PARI	He JAM	نق				AVı	455-6556	و	
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SHOP MANE, METAL PLATING AND WELDING	5	INC AND WE	5H 0 1 NG		100	BUILDING: 724			DATE	2 Mar 16	
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PASTE FLUX	1 3 1 2	0	0	0	0	9	0	0			
1429-FUBD 600	1212	0	0	0	0	0	9	0			
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ELECTRODE 6010	2 2	1/2 lb.	Lapered Lepened Governa Grand	2 thursd	Goors Gyouad	Exerca &	90	0,0	Trush	2 2	

DOVER AFB CHARACTERIZATION/HAZARDOUS MASTE MANAGEMENT SURVEY DOVER AFB DE (U) AIR FORCE OCCUPATIONAL AND ENVIRONMENTAL HEALTH LAB BROOKS AF R D BINOVI ET AL JUL 86 USAFOEHL-86-853E98652G1B F/G 13/2 2/2 AD-A178 785 UNCLASSIFIED NL



MICROCOPY RESOLUTION TEST CHART
NATIONAL BUREAU OF STANDARDS-1963-A

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DOVER AFB
SHOP HAME: METAL PLATING AND WELDING SHUP
SHOP SUPERVISOR: JYJSGT JOSEGSON

PAGE 2 OF 6 DATE:

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					DOVER AFB				PAGE	PAGE 3 OF 6	
SMOP HAME: METAL PLATING AND WELDING SHOP	ξ {	ING AND WE	LOING SHOP		BUILDING: 724	i: 724			DATE: 2 MAE	2 6	İ
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DATE: 2 Mar LE PAGE 4 OF S V. BUILDING: 724 DOVER AFB A95.410A SHOP NAME: METAL PLATING AND WELDING SHOP SHOP SUPERVISOR, MSGT

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3	LAYOUT FLUID 6850-00-684-9087	1										
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SHOP NAME, MACHINE SHOP	ي ع	Mer 10hite	110/12	الى.	BUILDING: 724	. 724		ā .¥	DATE: 24	RB	1
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A DIETHYLENE TAJAKINE	1										
TH 6010,000,005(4004)	1 2										
MEN PERQUIDE	1										
HI 8030-00-687-5624	2										
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M: PRESSURE GREASE 9150-PR-24837-8-1	1 2				' -							
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AUTOMOTIVE CREASE	1 2				 							
9150-00-180-0601	1 2				-							
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PAGE 2 OF 3 WSS -6.748	IS ANYTHING RINSED OR	1.8 drein (Y/N) 1.0 drein		100 M 200	146.4.4.9															
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Attachment 15
Acutel Toxicity to Daphnida of Various Chemicals Distribution List

ACUTE TOXICITY TO DAPHNIDS OF VARIOUS CHEMICALS

POLLUTANT	SPECIES	LEVEL	REF
ACENAPHTHENE	D.MAGNA	48HR LC50 41,200 ug/L	WPCF,80
ACROLEIN	D.MAGNA	48HR LC50 80 ug/L	WPCF,80
ACRYLONITRILE	D.MAGNA	48HR LC50 7550 ug/L	WPCF,80
ANTIMONY	D.MAGNA	48HR LC50 >530,000 ug/L	WPCF,80
ANTIMONY TRICHLORIDE	D.MAGNA	48HR LC50 19,000 ug/L	WPCF,80
ALACHLOR (LASSO)	D.PULEX	48HR EC50 3.8-12.3 mg/L	ECT,85
ARSENIC	DAPHNID	48HR LC50 1.7-3.8 mg/L	WPCF,85
ATRAZINE 4L	D.PULEX	48HR EC50 28.3-46.3 mg/L	ECT,85
BENZENE	D.MAGNA	48HR LC50 230000 ug/L	WPCF,80
BERYLLIUM	D.MAGNA	48HR LC50 2500 ug/L	WPCF,80
BIS-ETHER	D.MAGNA	48HR LC50 237000 ug/L	WPCF,80
BORIC ACID	D.MAGNA	48HR LC50 226 ug/L	WPCF,82
4BROMOPHENYL-PHENYL ETHER	D.MAGNA	48HR LC50 360 ug/L	WPCF,80
BROMOFORM	D.MAGNA	48HR LC50 46500 ug/L	WPCF,80
BUTYLBENZYL PHTHALATE	D.MAGNA	48HR LC50 92300 ug/L	WPCF,80
CADMIUM	DAPHNID	48HR LC50 24-118 ug/L	WPCF,85
CARBON TETRACHLORIDE	D.MAGNA	48HR LC50 35200 ug/L	WPCF,80
CHLOROBENZENE	D.MAGNA	48HR LC50 86000 ug/L	WPCF,80
CHLOROETHANOL	DAPHNID	48HR LC50 250-574 uL/L	WPCF,85
CHLOROFORM	D.MAGNA	48HR LC50 28900 ug/L	WPCF,80
4CHLORO-6METHYL PHENOL	D.MAGNA	48HR LC50 290 ug/L	WPCF,80
l, Chloronaphthalene	D.MAGNA	48HR LC50 1600 ug/L	WPCF,80
2, CHLOROPHENOL	D.MAGNA	48HR LC50 2580 ug/L	WPCF,80
4, CHLOROPHENOL	D.MAGNA	48HR LC50 4060 ug/L	WPCF,80
CHLORINE	D.MAGNA	1HR LC50 63 ug/L	WPCF,82
COPPER	DAPHNID	48HR LC50 17-57 ug/L	WPCF,85
1,2DICHLOROBENZENE	D.MAGNA	48HR LC50 2440 ug/L	WPCF,80
1,3DICHLOROBENZENE	D.MAGNA	48HR LC50 28100 ug/L	WPCF , 30

1.4010	LOROBENZENE	D.MAGNA	48HR	LC50	11000 ug/L	WPCF
CHLORDA	NE	D.MAGNA	MATC	16.2	ug/L	WPCF
1,2DICE	ILOROETHANE	D.MAGNA	48HR	LC50	218000 ug/L	WPCF
1,1DICE	ILOROETHENE	D.MAGNA	48HR	LC50	11600 ug/L	WPCF
2,4DICH	ILORO-6METHYLPHENOL	D.MAGNA	48HR	LC50	430 ug/L	WPCF
2,4DICE	ILOROPHENOL	D.MAGNA	48HR	LC50	2600 ug/L	WPCF
1,1DICE	ILOROPROPANE	D.MAGNA	48HR	LC50	23000 ug/L	WPCF
1,2DIC	ILOROPROPANE	D.MAGNA	48HR	LC50	52500 ug/L	WPCF
1,3DICH	LOROPROPANE	D.MAGNA	48HR	LC50	6150 ug/L	WPCF
CHROMIL	M	D.MAGNA	MATC	455	ug/L	WPCF
DIETHYL	PHTHALATE	D.MAGNA	48HR	LC50	52100 ug/L	WPCF
DI-2ETH	YL HEXYLPHTHALATE	D.MAGNA	48HR	LC50	11100 ug/L	WPCF
2,4DIME	THYLPHENOL	D.MAGNA	48HR	LC50	2120 ug/L	WPCF
DIMETHY	L PHTHALATE	D.MAGNA	48HR	LC50	33000 ug/L	WPCF
2,4DINI	TROPHENOL	D.MAGNA	48HR	LC50	4090 ug/L	WPCF
2,4DINI	TRO-6METHYLPHENOL	D.MAGNA	48HR	LC50	3120 ug/L	WPCF
2,3DINI	TROTOLUENE	D.MAGNA	48HR	LC50	660 ug/L	WPCF
1,2DIPH	ENYLHYDRAZINE	D.MAGNA	48HR	LC50	4100 ug/L	WPCF
DIELDRI	N	D.PULEX	48HR	EC50	251 ug/L	WPCF
DIMETHY	LQU I NONE	D.MAGNA	48HR	LC50	40 mg/L	WPCF
DINITRO	CRESOLS	D.MAGNA	48HR	LC50	33.4 mg/L	WPCF
ETHYLBE	NZENE	D.MAGNA	48HR	LC50	75000 ug/L	WPCF
FLUORAN	THENE	D.MAGNA	48HR	LC50	325000 ug/L	WPCF
FUELS (SOLUBLE FRACTION)					
NO.2	DIESEL	D.MAGNA	48HR	EC50	67,000 ppm	WPCF
NO.6	FUEL OIL	D. MAGNA	48HR	EC50	1,000,000 ppm	WPCF
FURADAN	4(CARBOFURAN)	D.PULEX	48HR	EC50	26.8-45.8 ug/L	ECT,
HEXACHL	OROETHANE	D.MAGNA	48HR	LC50	8070 ug/L	WPCF
ISOPHOR	ONE	D.MAGNA	48HR	LC50	117000 ug/L	WPCF
KEPONE		D.MAGNA	MATC	9-18	ug/L	WPCF

MERCURY	D. MAGNA	MATC 1.87 ug/L	WPCF,80
METHYLENE CHLORIDE	D.MAGNA	48HR LC50 224000 ug/L	WPCF,80
MIREX	D.MAGNA	MATC >34 ug/L	WPCF,82
NAPHTHALENE	D.MAGNA	48HR LC50 8570 ug/L	WPCF,80
NITROBENZENE	D.MAGNA	48HR LC50 27000 ug/L	WPCF,80
4NITROPHENOL	D.MAGNA	48HR LC50 21900 ug/L	WPCF,80
N-NITROSO-DIPHENYLAMINE	D.MAGNA	48HR LC50 7760 ug/L	WPCF,80
OCTACHLORONAPHTHALENE	D.MAGNA	48HR LC50 >530000 ug/L	WPCF,80
PENTACHLOROBENZENE	D.MAGNA	48HR LC50 5280 ug/L	WPCF,80
PENTACHLOROETHANE	D.MAGNA	48HR LC50 62900 ug/L	WPCF,80
PENTACHLOROPHENOL	DAPHNID	48HR LC50 140-280 ug/L	WPCF,85
PENTANEDIONE	DAPHNID	48HR LC50 35->50 uL/L	WPCF,85
PHENOL	D.MAGNA	48HR LC50 11800 ug/L	WPCF,80
SELENIOUS ACID	D.MAGNA	48HR LC50 1200 ug/L	WPCF,80
SELENIUM	D.MAGNA	48HR LC50 430 ug/L	WPCF,80
SILVER	DAPHNID	48HR LC50 11-15 ug/L	WPCF,85
SURFACTANTS	D.MAGNA	25DAY LC50 78-126 mg/L	WPCF,85
1,2,3,5TETRACHLOROBENZENE	D.MAGNA	48HR LC50 9710 ug/L	WPCF,80
1,2,4,5TETRACHLOROBENZENE	D.MAGNA	48HR LC50 >530000 ug/L	WPCF,80
1,1,1,2TETRACHLOROETHANE	D.MAGNA	48HR LC50 23900 ug/L	WPCF,80
1,1,2,2TETRACHLOROETHANE	D.MAGNA	48HR LC50 9320 ug/L	WPCF,80
TETRACHLOROETHENE	D.MAGNA	48HR LC50 17700 ug/L	WPCF,80
2,3,5,6TETRACHLOROPHENOL	D.MAGNA	48HR LC50 570 ug/L	WPCF,80
2,3,4,6TETRACHLOROPHENOL	D.MAGNA	48HR LC50 290 ug/L	WPCF,80
THALLIUM	D.MAGNA	48HR LC50 2180 ug/L	WPCF,80
TOLUENE	D.MAGNA	48HR LC50 313000 ug/L	WPCF,80
TOLUIDINES	D.MAGNA	48HR LC50 750 ug/L	WPCF,85
TOXAPHENE	D.MAGNA	48HR LC50 10 ug/L	WPCF,80
1,2,4TRICHLOROBENZENE	D.MAGNA	48HR LC50 50200 ug/L	WPCF,80
1,1,2TRICHLOROETHANE	D.MAGNA	48HR LC50 18000 ug/L	WPCF,80

1,1,1TRICHLOROETHANE	D.MAGNA	48HR LC50 >530000 ug/L	WPCF,80
TRICHLOROETHENE	D.MAGNA	48HR LC50 85200 ug/L	WPCF,80
2,4,5TRICHLOROPHENOL	D.MAGNA	48HR LC50 2660 ug/L	WPCF,80
2,4,6TRICHLOROPHENOL	D.MAGNA	48HR LC50 6040 ug/L	WPCF,80
2,4,6TRINITROPHENOL	D.MAGNA	48HR LC50 84700 ug/L	WPCF,80
ZINC	D.MAGNA D.MAGNA	48HR LC50 68-110ug/L 48HR EC50 1.1-1.7 mg/L	WPCF,85 WPCF,85

LC50 = median lethal concentration

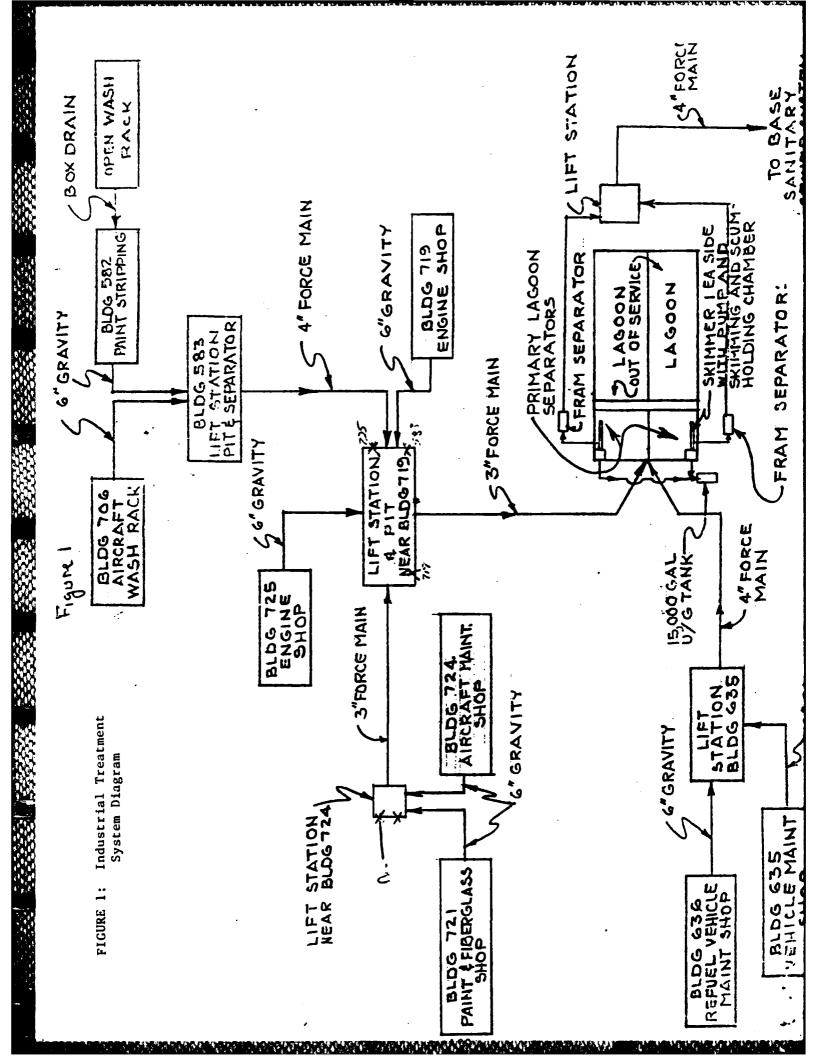
EC50 = median effective concentration

LD50 = median lethal dose

MATC = maximum acceptable toxicant concentration

WPCF = Journal Water Pollution Control Federation, June issue (literature review

ECT = Bulletin of Environmental Contamination Toxicology



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